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Karen Timberlake is Professor Emerita of Chemistry at Los Angeles Valley College, where she taught chemistry for allied health and preparatory chemistry for 36 years. She received her Bachelor's Degree in Chemistry from the University of Washington and her Master's Degree in Biochemistry from the University of California at Los Angeles.

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5 Given: mass = 50.0 g volume = 22.2 cm³ Plan: Place the mass and volume of the osmium metal in the density expression. $D = \frac{\text{mass}}{\text{volume}}$ $D = \frac{50.0 \text{ g}}{2.22 \text{ cm}^3}$ calculator = 22.522522 g/cm³ final answer (2) = 22.5 g/cm³ Solution

Chapter 1 Chemical Foundations 1.8 Density

A) Give the empirical formula of the following compound if a sample contains 0.0130 molC, 0.0390 molH, and 0.0065 molO B) Give the empirical formula of the following compound if a sample contains 11.66 g iron and 5.01 g oxygen

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