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series 5 2 (c) (ii) $K_a =$

$\frac{[CH_3COO^-][H^+]}{[CH_3COOH]}$

or with numbers $[H^+] =$

$\sqrt{(1.74 \times 10^{-5} \times$

$0.410)} = \sqrt{(7.13 \times 10^{-6})}$

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 $\text{pH} = 2.67 \times 10^{-3}$

Paper Jun 2018
 $\text{pH} = 2.57$ can give three

ticks here for (c)(ii)

penalise decimal

places < 2 > 1 1 1

allow HA etc here This

can be scored in

part(c)(i) but doesn't

CHEM4 Kinetics,

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