

## Chapter 4 Atomic Structure Test A Answers

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### Atomic Structure | Class 9 Chapter 4 Structure of Atom ...

CHAPTER 4 TEST: Atoms, Atomic Theory and Atomic Structure Matching. A. Bohr B. Democritus C. Rutherford D. Dalton E. Thomson F. Schrodinger \_\_\_\_ 1. Greek thinker; called nature's basic particle an atom, based on the Greek word "atomos" which means "indivisible". Did not have evidence that atoms existed. \_\_\_\_ 2. <http://www.brooklyn.k12.oh.us/userfiles/75/Classes/4711/NON-HONORS%20Chapter%204%20Test%202010.pdf>.

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Chapter 4: Atom (Test Answers) STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. chxistina. Key Concepts: Terms in this set (33) Dalton's atomic theory was accepted because a. Dalton showed how new molecules are formed. b. Dalton invented the electron microscope. c. Democritus said that it was correct.

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Section 4.2 Structure of the Nuclear Atom One change to Dalton's atomic theory is that atoms are divisible into subatomic particles: Electrons, protons, and neutrons are examples of these fundamental particles There are many other types of particles, but we will study these three

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Atomic Structure (Chapter Test A) STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. emily\_reidt. Terms in this set (21) the total number of protons and neutrons in the nucleus of an atom. mass number. the weighted average mass of the atoms in a naturally occurring sample of an element.

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Chapter 4 "Atomic Structure" Tools. Copy this to my account; E-mail to a friend; ... Help; Use these activities to help your study the vocabulary terms from this chapter. A B; atomic mass unit: defined as the mass of a carbon -12 atom: nucleus: central core of an atom, which contains most of the atom's mass: group: a vertical column of elements ...

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