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Answer Key Chapter 12: Stoichiometry
Mole Ratios Questions 1. Aluminum reacts with oxygen to produce aluminum oxide as follows: $4\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$
a. If you use 2.3 moles of Al, how many moles of Al_2O_3 can you make? b. If you want 3.9 moles of Al_2O_3 , how many moles of O_2 are needed? 2.

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and General Properties of the Metalloids;
18.4 Structure and General Properties of
the Nonmetals; 18.5 Occurrence,
Preparation, and Compounds of
Hydrogen; 18.6 Occurrence, Preparation,
and Properties of Carbonates; 18.7
Occurrence, Preparation, and Properties
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