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ΔH (IF) $-1 = -840$ kJ mol A. -190 B. -95 C.
 $+6$ D. $+95$ -1 7 2 5 7 -1 5 Which is a
correct definition of lattice enthalpy? A.
It is the enthalpy change that occurs
when an electron is removed from 1 mol
of gaseous atoms. B. It is the enthalpy

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change that occurs when 1 mol of a compound is formed from its elements.

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