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Solubility (metastable, at concentrations approaching saturation) also depends on the physical size of the crystal or droplet of solute (or, strictly speaking, on the specific surface area or molar surface area of the solute). For quantification, see the equation in the article on solubility equilibrium.For highly defective crystals, solubility may increase with the increasing degree of disorder.

Solubility - Wikipedia

As we noted at the end of Section 4.7, every solution x to the nonhomogeneous system $Ax = b$ can be written as $x_p + x_h$, where x_p is a particular solution to the given nonhomogeneous system, and x_h is a solution to the associated homogeneous system $Ax = 0$. Hence, the general solution to the given nonhomogeneous system has $n - k$ arbitrary ...

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