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In given process $dW = 0$, $dQ < 0$ then for a gas. 1. Temperature - increases 2. Volume - decreases. 3. Pressure - decreases. 4. Pressure - increases

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In the circuit given below, the reading of the voltmeter V will be .
1. 0. 2. 100 V. 3. 200 V. 4. 300 V

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The figure compares two solutions. Both solutions are represented by molecules, which consist of two small atoms attached to the bigger one. But solution on the left also has small positively charged atoms. Moreover, there are negatively charged molecules, which consist of one big atom connected to a small one in the solution on the right.

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