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Chapter 14 1. For the following reaction $K_c = 0.513$ at 500 K.
 $N_2O_4(g) \rightleftharpoons 2 NO_2(g)$ If a reaction vessel initially contains an N_2O_4 concentration of 0.0500 M at 500 K, what are the equilibrium concentrations of N_2O_4 and NO_2 at 500 K?

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