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Section 16.2 Heat and Thermodynamics (pages 479–483) This section discusses three kinds of thermal energy transfer and introduces the first, second, and third laws of thermodynamics. Reading Strategy (page 479) Build Vocabulary As you read this section, add definitions and examples to complete the table. For more information on this Reading

Chapter 16 Thermal Energy and Heat Section 16.2 Heat

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Section 16.2 Heat and Thermodynamics Pg. 479. Terms in this set (14) Thermal Energy. - the total energy of all the particles in an object. Thermal Energy transfers by: 1 - Conduction. 2 - Convection. 3 - Radiation. Conduction.

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16.2: Heat and Thermodynamics . Conduction •Conduction occurs within a material or between materials that are touching. •In conduction, collisions between particles transfer thermal energy, without any overall transfer of matter. •Metals are good thermal conductors.

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heat and thermodynamics of section 16.2: Heat and Thermodynamics; Conduction. The transfer of thermal energy with no overall transfer of matter is called conduction. The reason conduction is slower in gases than liquids and solids is because particles in gases collide less often.

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Section 16.2 Heat and thermodynamics Worksheet Answer Key

Section 16.2 – Heat and Thermodynamics Conduction is the transfer of thermal energy with no overall transfer of matter. Conduction occurs between materials that are touching. Conduction in gases is slower than in liquids and solids because the particles in a gas collide less often.

Chapter 16 - Thermal Energy and Heat

concluded that heat is not a form of matter. 474 Chapter 16 FOCUS Objectives 16.1.1 Explain how heat and work transfer

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energy. 16.1.2 Relate thermal energy to the motion of particles that make up a material. 16.1.3 Relate temperature to thermal energy and to thermal expansion. 16.1.4 Calculate thermal energy, temperature change, or mass

Section 16.1 16.1 Thermal Energy and Matter

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