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Section 2-3 Carbon Compounds. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. erinwd. carbon compounds flashcards. Terms in this set (76) What is organic chemistry the study of? the study of all compounds that contain bonds between carbon atoms. What is a monomer? small unit that can join together with other ...

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Section 2-3 Carbon Compounds This section explains how the element carbon able to form millions of carbon, or organic, compounds. It describes the four groups of organic compounds found in living things. The Chemistry of Carbon How many

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valence electrons does each carbon atom have? Each carbon atom has four electrons.

Section 2-3 Carbon Compounds

44 Chapter 2 1 FOCUS Objective 2.3.1 Describe the functions of each group of organic compounds. Vocabulary Preview As students read, have them make a concept map using the section's Vocabulary terms, excluding the words

Section 2-3 2-3 Carbon Compounds

Section 2.3 Carbon Compounds; Yara C. • 40 cards. Organic vs Inorganic Chemistry. Organic chemistry is the study of compounds that contain bonds between carbon atoms; Inorganic chemistry is the study of all other compounds; Importance of Carbon. 4 valence electrons allowing it to form strong ...

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Haseeb ...

Biology Section 2.3 Carbon Compounds. Test on Chapter 2 - 10/13/2015. STUDY. PLAY. What is the definition of organic and inorganic chemistry? organic chemistry - the study of compounds that contain bonds between carbon atoms inorganic chemistry - the study of all other compounds.

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Edl

Section 2-3 Carbon Compounds (pages 44-48) This section explains how the element carbon is able to form millions of carbon, or organic, compounds. It also describes the four groups of organic compounds found in living things. The Chemistry of

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Carbon (page 44) 1. How many valence electrons does each carbon atom have? Each carbon atom has four electrons. 2.

Macromolecules The Chemistry of Carbon

Chapter 2.3: Carbon Compounds (1) TRIGLYCERIDES - provide insulation that keeps you warm while protecting your internal organs with a layer of padding. (2) STEROIDS - include hormones and cholesterol. Cholesterol is produced by the body and consumed through food, and it... (3) PHOSPHOLIPIDS - ...

Chapter 2.3: Carbon Compounds Flashcards | Quizlet

Section 2.3 Exit Ticket 1. What properties of carbon explain carbon's ability to form different large and complex structures?
2. What are the four major categories of macromolecules? Describe the basic structures and primary functions of each. The End Title: Section 2.3 Carbon Compounds Author:

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Section 2.3 Carbon Compounds - WCS

2-3 Carbon Compounds A. The Chemistry of Carbon B. Macromolecules Mr. M. Varco St. Joseph High School • The study of “organic” chemistry involves the study of compounds containing bonds between carbon (C) atoms • Why is carbon worth studying? – Carbon atoms have four valence electrons allowing it to form strong covalent bonds

Chapter 2: The Chemistry of Life

Section 2-3: Carbon Compounds; Erica B. • 22 cards. Monomer; A small unit that can join with other small units to form polymers; Polymer; A large compound formed from combinations of many monomers; Carbohydrates. Compounds of carbon, hydrogen and oxygen. Ratio 1:2:1 ...

Section 2-3: CARBON COMPOUNDS - AP Bio with Kasuga at ...

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Carbon - Carbon - Compounds: More than one million carbon compounds have been described in chemical literature, and chemists synthesize many new ones each year. Much of the diversity and complexity of organic forms is due to the capacity of carbon atoms for bonding with one another in various chain and ring structures and three-dimensional conformations as well as for linking with other atoms.

Carbon - Compounds | Britannica

Section 2-3 Carbon Compounds Key Concept • What are the functions of each group of organic compounds? The Chemistry of Carbon(page 44) 1. How many valence electrons does each carbon atom have? 2. What gives carbon the ability to form chains that are almost unlimited in length? Macromolecules(page 45) 3.

Section 2-3 Carbon Compounds

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Section 2-3 Carbon Compounds (pages 44-48) ~ Key Concept •
What are the functions of each group of organic compounds?
The Chemistry of Carbon (page 44) 1. How many valence electrons does each carbon atom have? 2. What gives carbon the ability to form chains that are almost unlimited in length? Macromolecules (page 45) 3.

Section 2-3 Carbon Compounds (pages 44-48)

Carbon Compounds include that consist of which contain that consist of make up the structural units of RNA and DNA. building blocks of proteins which contain which contain which contain
Section 2-3 Concept Map Functions functions functions functions
Carbohydrates Lipids Nucleic acids Proteins Sugars and starches
Fats and oils Nucleotides Amino ...

Biology carbon compounds - LinkedIn SlideShare

Section 4: Observation 2: Compounds of Carbon and Hydrogen
Page 9/11

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Last updated; Save as PDF Page ID 75584; No headers. Many of the most important chemical fuels are compounds composed entirely of carbon and hydrogen, i.e. hydrocarbons. The smallest of these is methane (CH₄), a primary component of household natural gas.

Section 4: Observation 2: Compounds of Carbon and Hydrogen ...

Study Guide B Section 5: Enzymes Section Quiz 2.3: Carbon-Based Molecules Choose the letter of the best answer. _____ 1. Carbon is unique due to the carbon atom's a. bonding properties. b. six outer unpaired electrons. c. ionic compounds. d. hydrogen bonding strength. _____ 2. Which category of carbon-based molecules includes sugars and ...

SG 2.3 Carbon-based molecules - Weebly

Section 2-4 Chemical Reactions and Enzymes (pages 49-53) ...

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Chemicals Definition Reactants The elements or compounds that enter into a chemical reaction Products The elements or compounds produced by a chemical reaction 3. Chemical reactions always involve changes in chemical Energy in Reactions bonds . (page 50) Energy 5.

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