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The mole is a convenient unit for expressing the relative amounts of substances in chemical reactions. The burning of carbon in oxygen can be written with the 2 oxygen atoms bonded in a single O₂ molecule: The mole is the most common unit used to express the quantity of a chemical substance.

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What is a mole? The mole, commonly abbreviated mol, is the SI base unit used to measure the amount of a substance. It is the number of representat-ive particles, carbon atoms, in exactly 12 g of pure carbon-12. Through years of experimentation, it has been established that a mole of anything contains 6.022 136 7 10²³ representative particles. A representative particle is any

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Chapter 11: The Mole

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